Organic Chemistry II – 1323 CHEM 2452 Section 1 Summer Quarter, 2019



Welcome to Organic Chemistry II! This is the second of a three-quarter series in organic chemistry. The scope of this course broadly focuses on the chemistry of carbon and its compounds. We will discuss basic principles regarding chemical bonding, structure, and classification of organic molecules. We will then apply these concepts to study the chemical reactivity of such compounds. Understanding how molecules interact with others will allow for appreciation of chemical synthesis for the production of compounds useful for society, including pharmaceuticals, agrochemicals, plastics, pesticides and other materials.

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Office: Seeley G. Mudd, Room 132

Lecture and Recitation: MTWRF 9:20 a.m. – 11:20 a.m. in Boettcher Center Auditorium, Room 103

Office Hours: By appointment. Please e-mail me at least one day in advance.

Textbook: Organic Chemistry, Sixth Edition, by Marc Loudon and Jim Parise (Required)

Study Guide: *Study Guide and Solutions Manual to Accompany, Organic Chemistry, Sixth Edition,* by Marc Loudon and Jim Parise (Required)

Molecular Models: Molymod #62053 Organic Chemistry Molecular Model Set by Indigo Instruments or Darling Models from DU Bookstore (Required)

Exams: There will be two Friday exams during the quarter worth 100 points each. The final exam will also be worth 100 points. If your final exam score is higher than any exam score, the lowest score will be dropped and the final will count for 200 points. *There will be no makeup exams. If you miss an exam for any reason, that exam will be dropped and the final will count for 200 points. The final exam is not optional.*

Final Grade: Your final letter grade will be determined out of 300 points and will be curved appropriately based on overall class performance.

Lectures: I will cover most material on the white board using three different colors. I would recommend bringing at least 3 colored pens/pencils to class. *I will not post my lecture notes online. If you miss a lecture, please see a classmate for the notes.* If PowerPoint slides are periodically incorporated in lecture they will be posted afterwards on Canvas.

Canvas: The University of Denver uses Canvas as its learning management system. You may log in to <u>https://du.instructure.com</u> with your DU ID number and PioneerWeb password to access the course. Please ensure your settings allow for e-mail announcement notifications. Here are some helpful Canvas resources to get you started:

Canvas Student Quickstart Guide: <u>http://guides.instructure.com/m/8470</u> Canvas Student Guide: <u>http://guides.instructure.com/m/4212</u>

Academic Integrity: I have high expectations for each and every one of you as students at the University of Denver. While I encourage group study sessions outside of class, I expect you to work independently during in class examinations. Any deviations from this policy will not be tolerated. For more information, please see the University of Denver's official Honor Code at: http://www.du.edu/studentlife/studentconduct/

Science and Engineering Center: Need extra help? The Science and Engineering Learning Center is a collaborative space staffed by undergraduate and graduate learning assistants (LAs) trained to assist students with some first and second year biology, chemistry, physics, computer science and engineering courses. We offer support for both lecture and laboratory courses for chemistry, physics, and engineering courses and lecture only for computer science and biology. Our goal is to help students grow as problem solvers by assisting with homework sets, lab reports, and preparing for exams. The Science and Engineering Learning Center is **not** a one-on-one tutoring center, but is rather a support system where students can get guidance from LAs as well as their peers. This center is open to all DU students. All services are free. Located in the north-west corner of the first floor of the Anderson Academic Commons (west of the writing center). See http://portfolio.du.edu/sec for a complete schedule.

Week #: Start	Date Topic	Reading
1 : 07/08/19	Chapter 10: Alcohols and Thiols Chapter 10: Oxidation and Chemical Synthesis	рр 452 — 476 рр 476 — 504
	Chapter 11: Ethers, Epoxides, Glycols, Sulfides Chapter 11: <i>Continued</i>	рр 511 — 536 рр 536 — 560
	Chapter 12: Introduction to Spectroscopy Chapter 12: <i>Continued</i> EXAMINATION 1 (material from Chapter 10.1 through 12.6)	рр 569 — 585 рр 585 — 605
2 : 07/15/19	Chapter 13: NMR Spectroscopy Chapter 13: <i>Continued</i>	рр 611 — 645 рр 645 — 672
	Chapter 14: Alkynes	pp 681 – 708
	Chapter 15: Dienes, Resonance, Aromaticity Chapter 15: <i>Continued</i>	рр 712 – 741 рр 741 – 780
	Chapter 16: Benzene and Derivatives Chapter 16: Electrophilic Aromatic Substitution EXAMINATION 2 (material from Chapter 10.1 through 16.3)	рр 789 — 810 рр 810 — 830
3 : 07/22/19	Chapter 17: Allylic and Benzylic Reactivity Chapter 17: <i>Continued</i>	рр 836 – 851 рр 851 – 870
	Chapter 18: Aryl Halides Chapter 18: <i>Continued</i>	рр 879 – 901 рр 901 – 935
	Finish Course Material and Review	

07/26/19 (F) FINAL EXAMINATION (material from Chapter 10.1 through 18.11) 9:20 – 11:20 am