

**Organic Chemistry III – 4381  
CHEM 2453 Section 2  
Spring Quarter, 2018**



**Instructor:** Asst. Professor Brian W. Michel

**E-mail:** [Brian.Michel@du.edu](mailto:Brian.Michel@du.edu)

**Phone:** 303.871.2595

**Office:** Seeley G. Mudd 232

**Lecture:** TR 8:00 am – 9:50 am (Olin 105)

**Office Hours:** By appointment, please email. I am very happy to meet with students and will try to respond quickly to set up a time when we can meet. Don't be shy, come say hi!

**Required Text:** *Organic Chemistry, Sixth Edition*, by Marc Loudon and *Study Guide and Solutions Manual to Accompany Organic Chemistry, 6<sup>th</sup> Edition*, by Marc Loudon and Joseph G. Stowell

**Recommended Model Kit:** Molymod #62053 Organic Chemistry Molecular Model Set by Indigo Instruments or other similar kit (typically available on Amazon)

**Course Objective:** In this third and final quarter of the three quarter organic chemistry series we will continue to learn about the ways in which elements bond together to form molecules (especially those containing carbon). We will continue to build on the understanding of bonds (made up of electrons), reactions (breaking/making bonds, i.e. the flow of electrons) and methods for determining structure (NMR, IR, MS). We will learn about a number of new functional groups, many of which play prominent roles in the chemistry in biological systems. As before, we will learn about the conformation and three-dimensional structures of molecules and how these considerations influence their reactivity. Most importantly, it is my goal that all of you will not memorize the facts of organic chemistry, but come to understand the “why” of organic chemistry. Nearly everything in organic chemistry builds upon previous knowledge and is highly interrelated; therefore it is important to study for understanding and not just memorization. It is important to relate new information to the topics we previously covered, assign relative values and begin to perceive patterns. When you understand why atoms bond together, why structures are more or less favorable, and why molecules react in the ways that they do you will be able to solve challenging problems that are beyond

the scope of what you have previously seen. Learning to develop problem solving skills in this way will translate beyond this course and even other chemistry classes.

**Lectures:** The lectures will generally follow the progression of the textbook. Most lectures will be presented on the board. ***Attending every class and taking meaningful notes is incredibly important for this complex subject.*** Staying up with the reading will help you to understand the lecture better and take more meaningful notes. Both during the lectures and the recitation session we will work through problems. Working problems will be a pillar of your success in organic chemistry and by working through these together I hope to provide you with insight from my experiences, techniques for reasoning and problem solving skills.

**Recitation:** There is no recitation in this section of organic chemistry III, however we will have time to work through problems in class and there

Wednesday recitations will provide us with an opportunity to go over challenging problems from the homework assignments or review particular topics. We may use some recitation sections for a normal lecture. There may also be short quizzes (1-2 questions, ~20 min) that could be worth bonus points...

**Homework:** Working as many problems as possible is crucial to successfully reinforcing your understanding and building problem solving skills that will allow you to be successful on examinations.

- **Online Homework through Sapling:** There will be weekly problem sets to be completed online through the Sapling system. **Homework will be due Monday nights at 11 pm.** These problems will be graded and normalized to 50 points over the course of the quarter. The lowest weekly homework score will be dropped. *Late submissions receive no credit.*
  - Go to [www.saplinglearning.com/login](http://www.saplinglearning.com/login) to log in or create an account.
  - Under Enroll in a new course, you should see Courses at University of Denver. Click to expand this list and see courses arranged by subject. Click on a subject to see the terms that courses are available.
  - Click on the term to expand the menu further (note that Semester 1 refers to the first course in a sequence and not necessarily the first term of the school year).
  - Once the menus are fully expanded, you'll see a link to a specific course. If this is indeed the course you'd like to register for, click the link.
  - Review the [system requirements](#) and confirm that Flash is updated and enabled in your browser.
- **From the text:** I will assign additional problems from the text and select problems will be graded for bonus points. In addition to these being valuable in building your problem solving skills and providing additional practice, exam problems can be very similar or variants of these homework problems. While the answers to many problems are provided in the study guide, you should not consult the answer key until you have struggled to solve the problem on your own. In this way you will help yourself to identify areas where you may be making incorrect assumptions or misunderstanding a concept.

Working problems together can also be very helpful and I encourage you to go over solutions to problems in small groups after working the problems independently. Your peers may have an alternate way of looking at a problem than you or I might, which can add to your toolbox of problem solving skills. Additionally, helping to teach the subject to your classmates is amongst the best ways to understand a concept yourself.

**Exams:** There will be two midterm exams during the quarter worth 100 points each and a final exam also worth 100 points. If your final exam score is higher than either midterm exam score, the lowest score will be dropped and the final will count for 200 points.

**Final Grade:** Your final grade will be determined out of the 350 available points on exams and homework (plus all earned bonus points) and will be appropriately curved based on the overall class performance. **There will be no makeup exams. If you miss an exam for any reason, that exam will be dropped and the final will count for 200 points. The final exam is not optional.**

#### **Cell Phone and Electronic Device Policy:**

While I understand that mobile devices have become integral to our lives, they are disruptive to the learning environment. Therefore, I request that all electronic devices be turned off (not muted) during class (i.e.; please don't text/facebook during class). If an emergency arises, and you need to contact the outside world during our lecture or recitation time, I request that you quietly leave the room and conduct your conversation outside. Additionally, most all of our lectures will require far too much structural drawing for effective notes to be taken on a laptop so please leave these devices off during lecture.

#### **Lecture and Testing Accommodations:**

I will make every effort to accommodate students diagnosed with a learning disability. I will do this in complete confidence. I do, however, request that any student requiring these accommodations inform me the first week of class. For further information, please see the University Disability Services' website at <http://www.du.edu/disability/dsp/index.html>.

#### **Academic Integrity:**

While I advocate collaborative learning and teamwork, I also firmly believe that each individual should maintain the highest ethical standards. As such, I support and will strictly enforce the Honor Code of the University of Denver. [www.du.edu/honorcode](http://www.du.edu/honorcode).

#### **Honor Code Statement.**

All members of the University of Denver are expected to uphold the values of *Integrity*, *Respect*, and *Responsibility*. These values embody the standards of conduct for students, staff, faculty, and administrators as members of the University community. These values are defined as:

*Integrity:* acting in an honest and ethical manner;

*Respect:* honoring differences in people, ideas, and opinions;

*Responsibility:* accepting ownership for one's own conduct.

**Pioneer Pledge.**

As a University of Denver Pioneer I pledge...

- to act with INTEGRITY and pursue academic excellence;
- to RESPECT differences in people, ideas, and opinions and;
- to accept my RESPONSIBILITY as a local and global citizen;

Because I take pride in the University of Denver I will uphold the *Honor Code* and encourage others to follow my example.

**Topics to be covered:** Preliminary Course Schedule – Subject to Change

<b>Date</b>	<b>Topic</b>	<b>Reading</b>
3/27/18	Chapter 18: Aryl Halides	
3/29/18	Chapter 18: Transition Metal Catalysis	
4/3/18	Chapter 19: Aldehydes and Ketones	
4/5/18	Chapter 19: Continued	
4/10/18	Chapter 20: Carboxylic Acids	
4/12/18	Chapter 20: Continued	
4/17/18	Catch up and Review	
4/19/18	<b>Exam 1</b>	
4/24/18	Chapter 21: Carboxylic Acid derivatives	
4/26/18	Chapter 21: Continued	
5/1/18	Chapter 22: Enolates, enols, $\alpha,\beta$ -unsaturated compounds	
5/3/18	Chapter 22: Continued	
5/8/18	Chapter 23: Amines	
5/10/18	Chapter 23: Continued	
5/15/18	Catch up and Review	
5/17/18	<b>Exam 2</b>	
5/22/18	Chapter 24: Carbohydrates	
5/24/18	Chapter 25: Select section	
5/29/18	Chapter 26: Select sections	
5/31/18	Chapter 27: Select sections	
6/5/18	<b>Final Exam</b>	

**Canvas and Class Notes:**

For most lectures information will be presented on the board, however, when PowerPoint slides are used they will be posted on Canvas. Suggested problems and additional practice problems will be posted on Canvas.