# CHEM 3702 Special Topics, Analytical Mass Spectrometry; Principles and Applications Fall 2004

Instructor

Dr. Verl Murugaverl

## **Objectives:**

This course objective is to make you familiar with modern mass spectrometry and its applications to a wide range of studies. This course will provide an advanced training in both theoretical and practical aspects of MS to those who are planning on using state-of-the-art mass spectrometers. The topics include ionization techniques, mass analyzers, fragmentation, tandem mass spectrometry, and more specific areas like proteomics, post-translational modifications and principles of coupling of separation techniques with MS.

Experiments will be carried out to demonstrate the utility of MALDI/TOF, ESI/ITMS in macromolecular analysis and the use of LC and GC coupled with MS.

# Prerequisites:

Sound knowledge of organic chemistry will be very helpful. You will be responsible for searching the literature on the topics covered (in the lectures) for additional (in depth) information.

#### **Evaluation:**

Your comprehension of the material covered will be evaluated by three 30 minutes quizzes and your experimental skills.

The participants will be working in groups on various problems including developing strategies to solve analytical problems and executing the experiments. Each student is expected to present a short report on their investigations at the end of the course.

#### Literature:

Any texts on Mass Spectrometry and its applications. Review Articles Scientific Publications

### Useful web sites:

<u>www.ms.mc.vanderbilt.edu/tutorials/ms</u> <u>www.methods.ch.cam.ac.uk/meth/ms</u>

# **Environmental Chemistry 1: Atmospheric Chemistry**

Don Stedman 303 871-2580 Home 303 757-2489 and talk to the machine

Problem sets may be done as collaboratively as you wish, but each student must turn one in.

Late problem sets lose 25% of their value per day late Office Hours any noon hour: SGM 101 or by appointment

Sept		
13	Composition of the atmosphere and combustion stoichiom	netry
15	equilibrium products of combustion	ich y
17	coal, gasoline and diesel	
, 17	coal, gasoline and dieser	
20	Kinetics and departure from equilibrium	
22	Zeldovich NO formation	
24	Atmospheric fate of NO and SO2	problem set 1 (stoichiometry) due
		, ,
27	Air pollution general	
29	London	
1-Oct	Los Angeles and Denver	discuss Holzworth
		If I can find it
4	NO2 photochemistry 1 NOx and Ox	
6	NO2 photochemistry 2 VOC NMHC and HOx	
8	Beer's law photochemistry and visibility	problem set 2 (Kp) due
	•	(4)
11	Regional air pollution	
13	geochemical cycles and steady state	
15	MID TERM	problem set 3 (kinetics) due
	·	,
X 18	Particle formation in the atmosphere SL	
20	Particle formation in combustion SL	·
22	Particles and visibility SL	
25	Formation of atmospharia layers Cools hairly	
25 27	Formation of atmospheric layers. Scale height	
21 29	ionosphere	
	stratosphere	Particle problem set 1 due
Nov		
1	Chapman mechanism	
3	HOx NOx ClOx etc	
5	spray cans and the ozone layer	
8	Polar ozone hole	
10	control techniques laws and technologies	
12	control techniques laws and technologies	Particle problem set 2 due
	22	i article problem set 2 due
x 15	Particles and lungs SL	
x 17	Particles and climate SL	
x 19	Particles in the stratosphere SL	
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22	8:00 - 10:00 am FINAL EXAM	