Organic Chemistry I – 2271 CHEM 2451 Section 2 Winter Quarter, 2013

Welcome to Organic Chemistry I! This is the first of a three-quarter series in organic chemistry. The scope of this course broadly focuses on the chemistry of carbon and its compounds. We will discuss basic principles regarding chemical bonding, structure, and classification of organic molecules. We will then apply these concepts to study the chemical reactivity of such compounds. Understanding how molecules interact with others will allow for appreciation of chemical synthesis for the production of compounds useful for society, including pharmaceuticals, agrochemicals, plastics, pesticides and other materials.

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Office: Seeley G. Mudd, Room 232

Lecture: T/R 8:30 a.m. – 9:50 a.m.

Recitation: W 8:00 a.m. – 8:50 a.m.

Office Hours: By appointment. Please e-mail me at least one day in advance.

Textbook: Organic Chemistry, Fifth Edition, by Marc Loudon (Required)

Study Guide: Study Guide and Solutions Manual to Accompany, Organic Chemistry, Fifth Edition, by Marc Loudon and Joseph G. Stowell (Required)

Molecular Models: Molymod #62053 Organic Chemistry Molecular Model Set by Indigo Instruments (Required)

Exams: There will be two 60-minute exams during the quarter worth 100 points each. The final exam will be worth 200 points. If your final exam score is higher than either 60-minute exam score, the lowest score will be dropped and the final will count for 300 points. There will be no makeup exams. If you miss an exam for any reason, that exam will be dropped and the final will count for 300 points. The final exam is not optional.

Online Homework through Sapling: There will be weekly problem sets to be completed online through the Sapling system. These problems are intended to help you understand the lecture material and reading assignments more thoroughly. We will review the problems during Wednesday morning recitation

sections. Homework will be due Tuesday nights by 11 p.m. The online problems will be graded and are worth 50 *total points* for the course. The lowest weekly homework score will be dropped. Late submissions receive no credit.

Final Grade: Your final letter grade will be determined out of 450 points and will be curved appropriately based on overall class performance.

Lectures: I will cover most material on the white board using up to three different colors. I would recommend bringing at least 3 colored pens/pencils to class. *I will not post my lecture notes online. If you miss a lecture, please see a classmate for the notes.* If PowerPoint slides are periodically incorporated in lecture they will be posted afterwards on Blackboard.

Date	Торіс	Reading
01/08/13	Chapter 1: Introduction	pp 01 – 29
01/09/13	Chapter 1: Chemical Bonding and Structure	pp 29 – 43
01/10/13	Chapter 2: Chemistry of Alkanes	pp 46 – 67
01/15/13	Chapter 2: Continued	pp 67 – 83
01/17/13	Chapter 3: Curved-Arrow Notation	pp 87 – 96
01/22/13	Chapter 3: Acids and Bases	pp 96 – 117
01/24/13	Chapter 4: Structure and Reactivity of Alkenes	pp 122 – 147
01/29/13	Chapter 4: Continued	pp 147 – 174
01/31/13	EXAMINATION I (material from Chapter 1.1 throug	jh 4.5)
02/05/13	Chapter 5: Addition Reactions of Alkenes	pp 178 – 196
02/07/13	Chapter 5: Continued	pp 196 – 220
02/12/13	Chapter 6: Principles of Stereochemistry	pp 226 – 246
02/14/13	Chapter 6: Continued	pp 246 – 263
02/19/13	Chapter 7: Conformational Analysis	pp 268 – 290
02/21/13	Chapter 7: Stereochemistry of Reactions	pp 290 – 316

Preliminary Course Schedule – Subject to Change

02/26/13	Chapter 8: Alkyl Halides and Chalcogens	pp 323 – 355
02/28/13	EXAMINATION II (material from Chapter 1.1 through	gh 7.9)
03/05/13	Chapter 8: Continued	pp 355 – 373
03/07/13	Chapter 9: Reactions of Alkyl Halides	pp 377 – 400
03/12/13	Chapter 9: Continued	pp 400 – 429

03/14/13 FINAL EXAMINATION (material from Chapter 1.1 through 9.8)



Instructions for Using Sapling: Sapling's chemistry questions are delivered in a web browser to provide real-time grading, response-specific coaching, improvement of problem-solving skills, and detailed answer explanations. Dynamic answer modules enable one to interact with 3D models and figures, utilize drag-and-drop synthetic routes, and draw chemical structures - including stereochemistry and curved arrows. Altogether, Sapling is cheaper than a tutor, provides more value than a solutions manual, and goes beyond a mere assessment exercise to give a learning experience.

Students:

1. Go to http://saplinglearning.com

2a. If you already have a Sapling Learning account, log in then skip to step 3. 2b. If you have Facebook account, you can use it to quickly create a SaplingLearning account. Click the blue button with the Facebook symbol on it (just to the left of the username field). The form will auto-fill with information from your Facebook account (you may need to log into Facebook in the popup window first). Choose a password and timezone, accept the site policy agreement, and click "Create my new account". You can then skip to step 3.

2c. Otherwise, click "Register here". Supply the requested information and click "Create my new account". Check your email (and spam filter) for a message from Sapling Learning and click on the link provided in that email.

3. Find your course in the list (you may need to expand the subject and term categories) and click the link.

4. Select a payment option (or bookstore) and follow the remaining instructions.

Once you have registered and enrolled, you can log in at any time to complete or review your homework assignments. During sign up, and throughout the term, if you have any technical problems or grading issues, send an email to <u>support@saplinglearning.com</u> explaining the issue. The Sapling support team is almost always more able (and faster) to resolve issues than your instructor.