

# Our Biosphere: Past and Future

Music to gather by

**Academy for Lifelong Learning**

October 24 - November 14, 2017

Jonathan F. Ormes

[JFOrmes@comcast.net](mailto:JFOrmes@comcast.net)

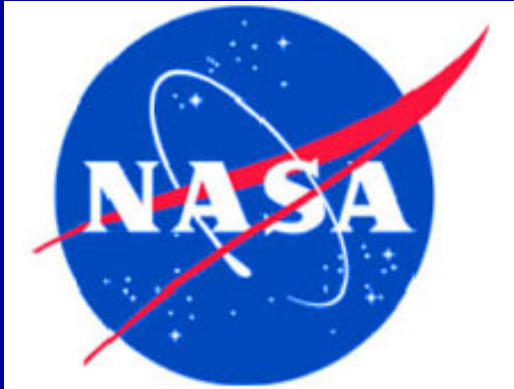
## Stanford University



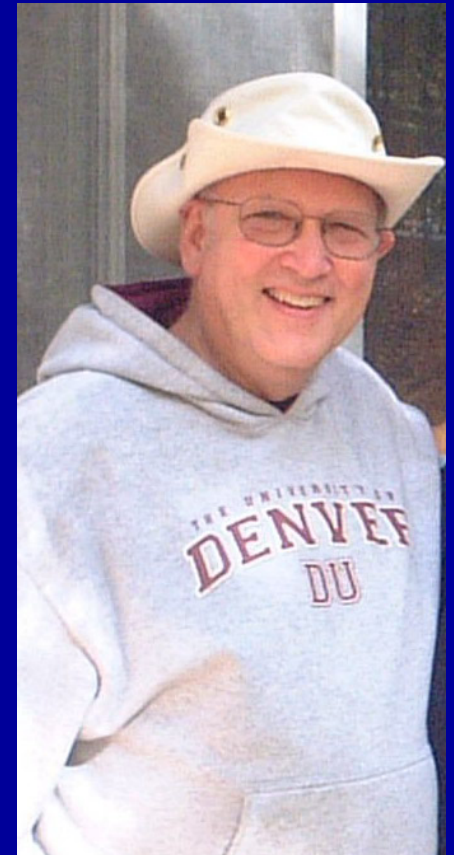
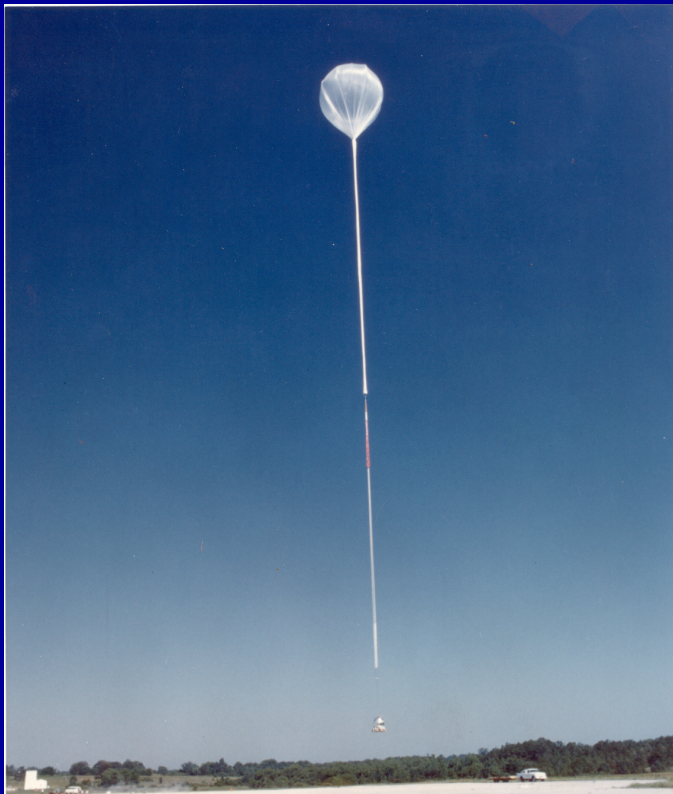
UNIVERSITY OF MINNESOTA







## Director of Space Sciences 2000-2004



# Public Announcement

Presentation slides and handouts can be downloaded from either the link at the Academy's web site or directly from this Web site at DU:

<http://portfolio.du.edu/OurClimate>

The site is public.  
You do not need to create an account.  
Search for "[OurClimate](#)".

The Academy also has a link to my page there.  
Click on **Courses** and then on the last item on the scroll down menu  
**Course Materials**

<http://www.academyll.org/course-materials/>



# Life might be called the “music of the spheres”

## Play this

- Lithosphere
  - Rocky
- Atmosphere
  - Vapor
- Hydrosphere
  - Water
- Biosphere
  - Life

**Boundaries are sometimes “fuzzy”**

**Hydro can include water in clouds and on land**

**Biota are found in all the other spheres, even deep underground extremophiles**

**Interesting things happen at the boundaries**

Earth's biosphere temperature is controlled  
by a very thin layer of atmosphere!



The Greenhouse Effect keeps the planet warm and habitable.



# Course outline

## 1. Origin of the Biosphere and Life

- Planet Formation and the Origin of Life
- Life on Earth: By Chance or By Law?

## 2. Guest speaker: Alexandra Jahn

- Paleoclimate
- Assistant Professor, Dept. for Atmospheric and Oceanic Sciences, CU-Boulder

## 3. Energy revolution and climate prospects

## 4. Predicting the future

- Ethics and scarcity



***Garden of Earthly Delights***  
by Hieronymus Bosch

How to describe the indescribable

"But I don't want to go among mad people," Alice remarked.

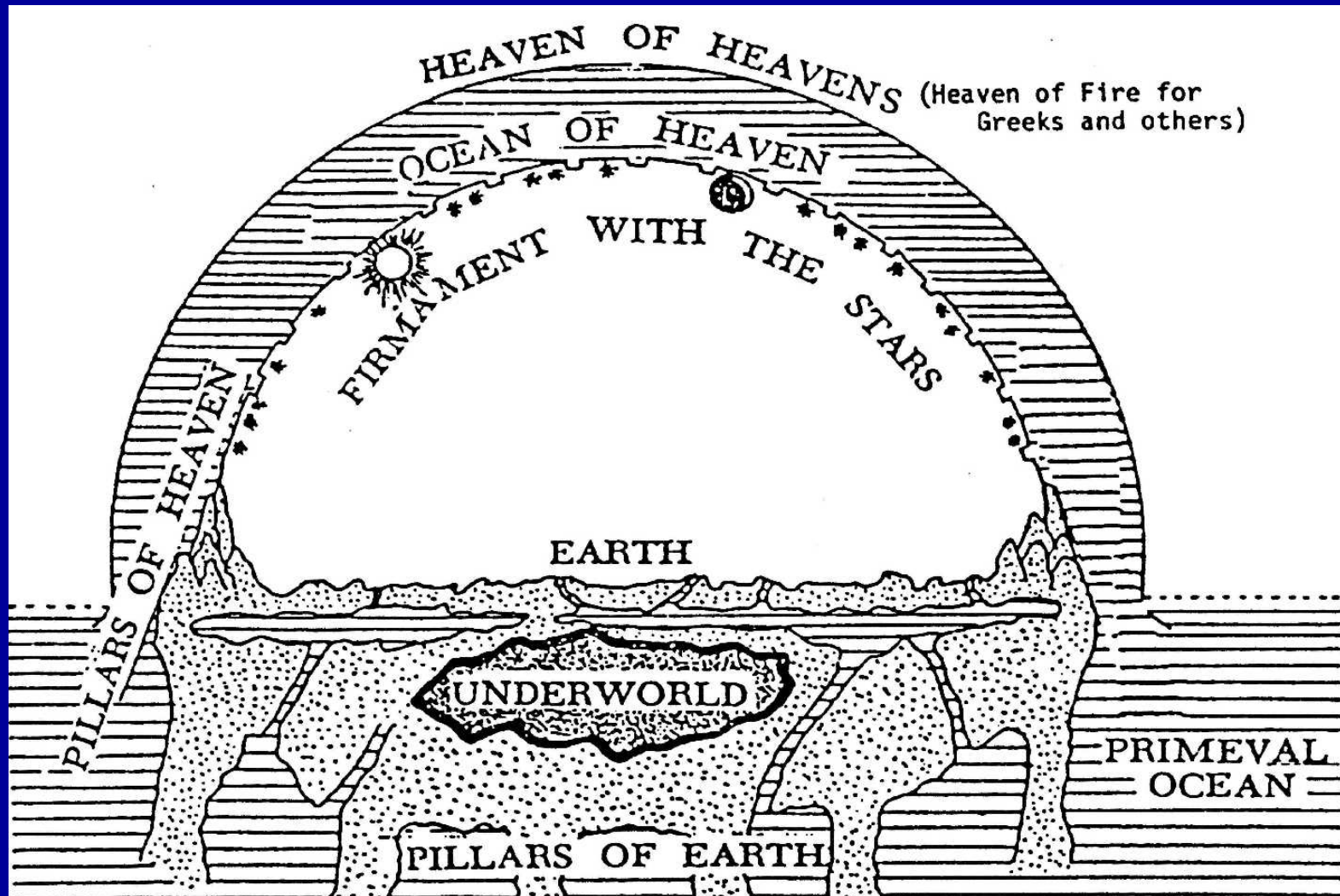
"Oh, you can't help that," said the Cat: "we're all mad here. I'm mad. You're mad."

"How do you know I'm mad?" said Alice.

"You must be," said the Cat, "or you wouldn't have come here."

—Lewis Carroll, *Alice in Wonderland*





<http://rosarubicondior.blogspot.com/2013/01/why-god-couldnt-have-written-bible.html>

"I'll say one thing for the Bible: it's full of good laughs."

Rosa Rubicondior

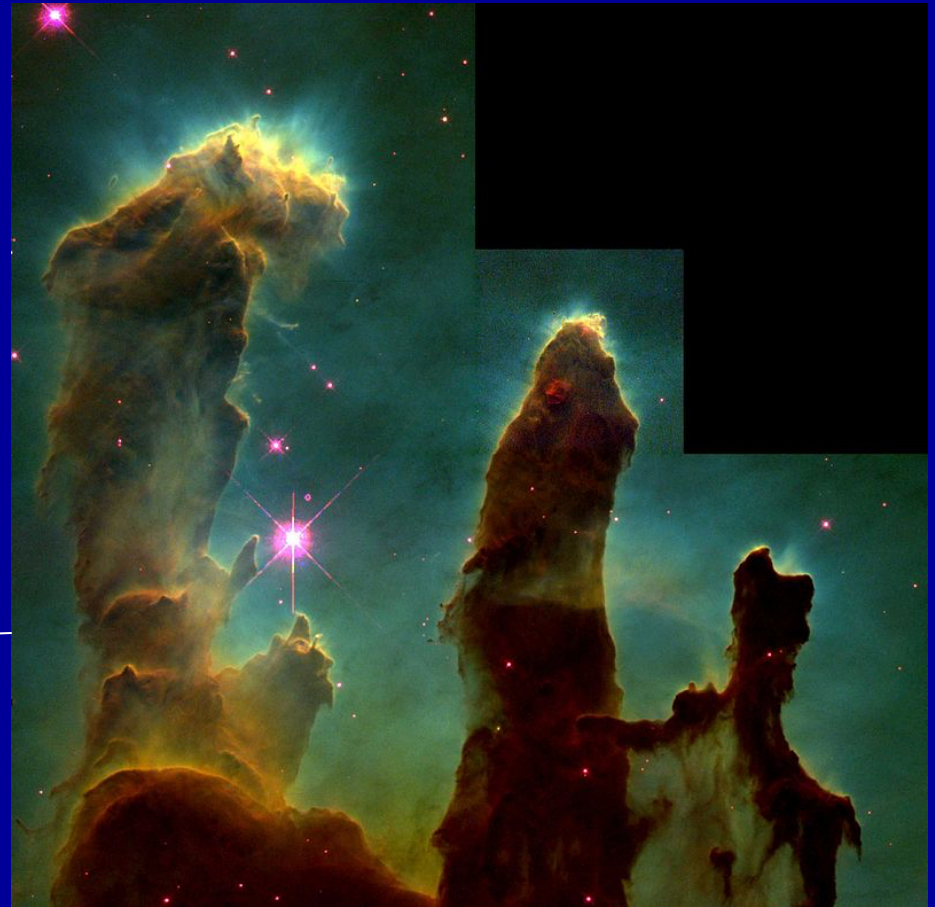
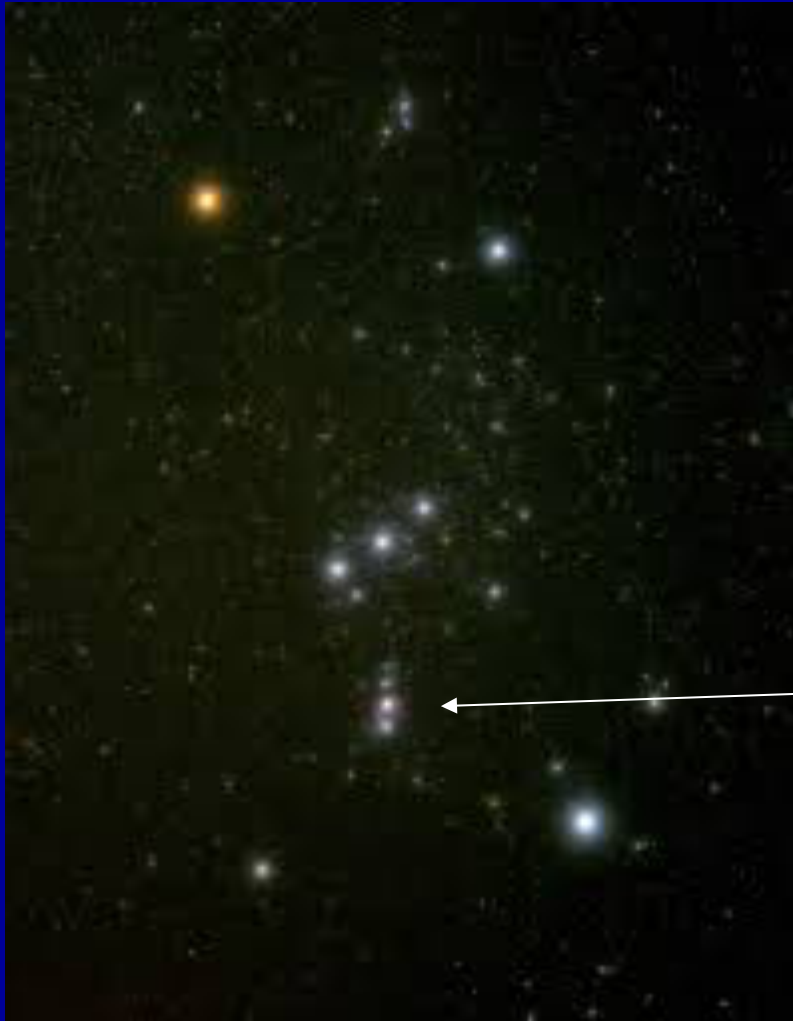
The Flammarion engraving (1888), so called because it first appeared in his book *L'atmosphère: météorologie populaire* ("The Atmosphere: Popular Meteorology")



The engraving depicts a traveler who arrives at the edge of a flat Earth and sticks his head through the firmament.

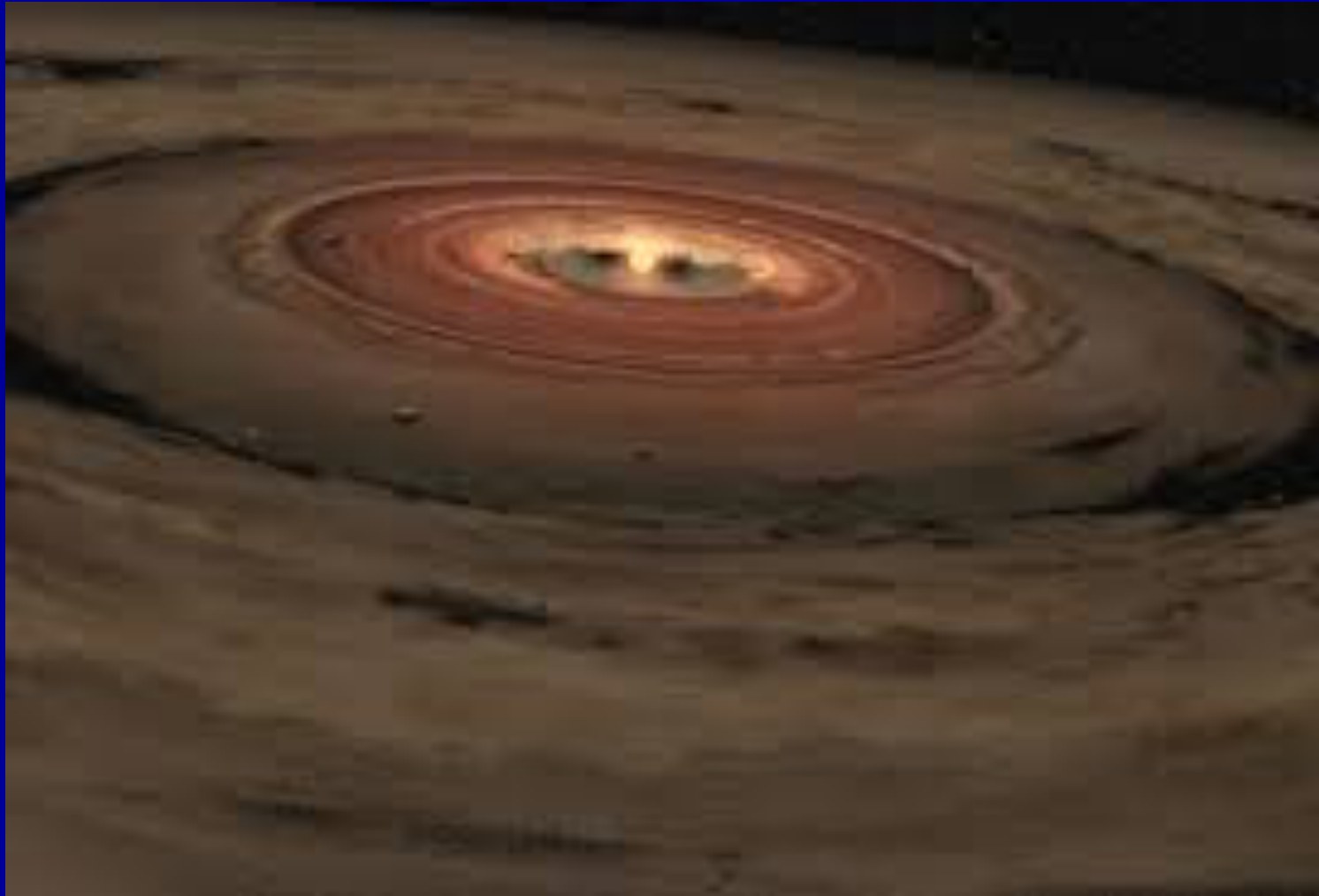


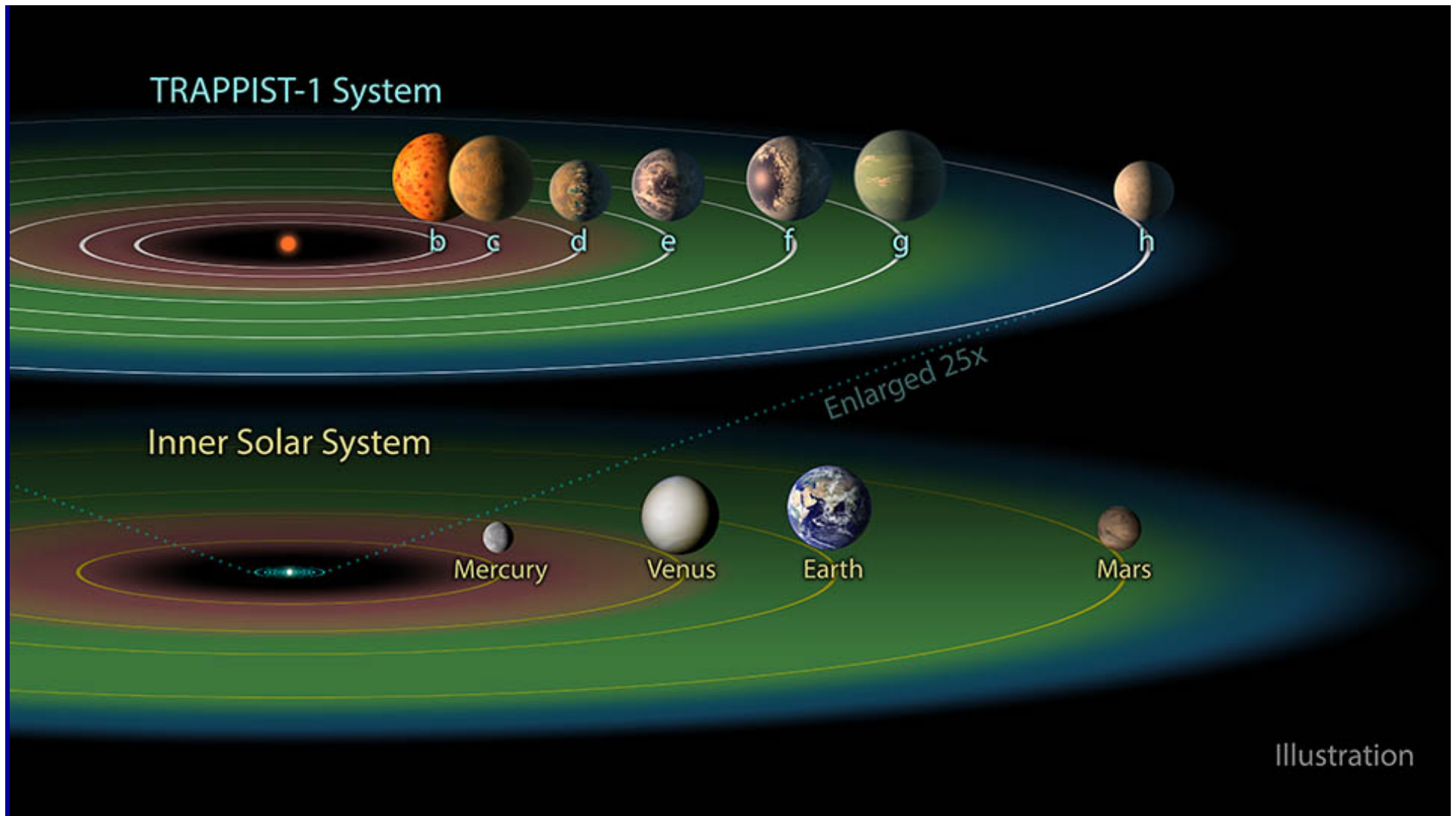
# Orion Nebula



**Credit: Hubble Space Telescope**

**protostar**





Illustration

**Kepler: 1200 planets =>  
500 million planets orbiting stars like sun in our Milky Way galaxy**



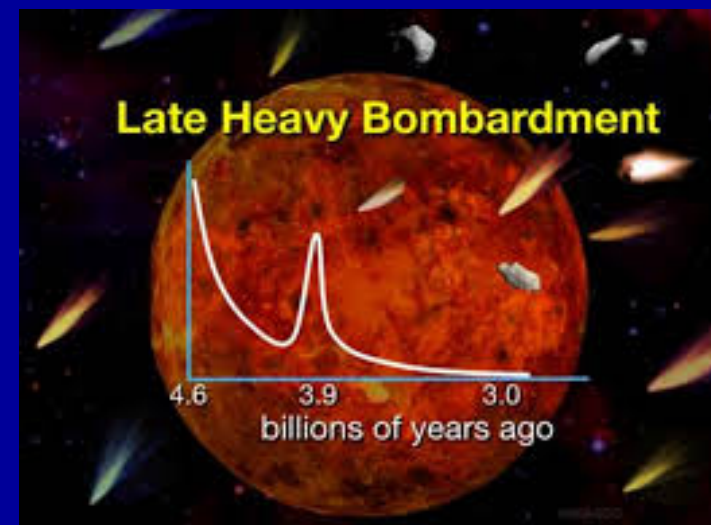




**Collision with  
another fairly large  
“planet”**

**Formed the moon.**

**Data from the study of  
impact craters on the moon.**







Source: NASA

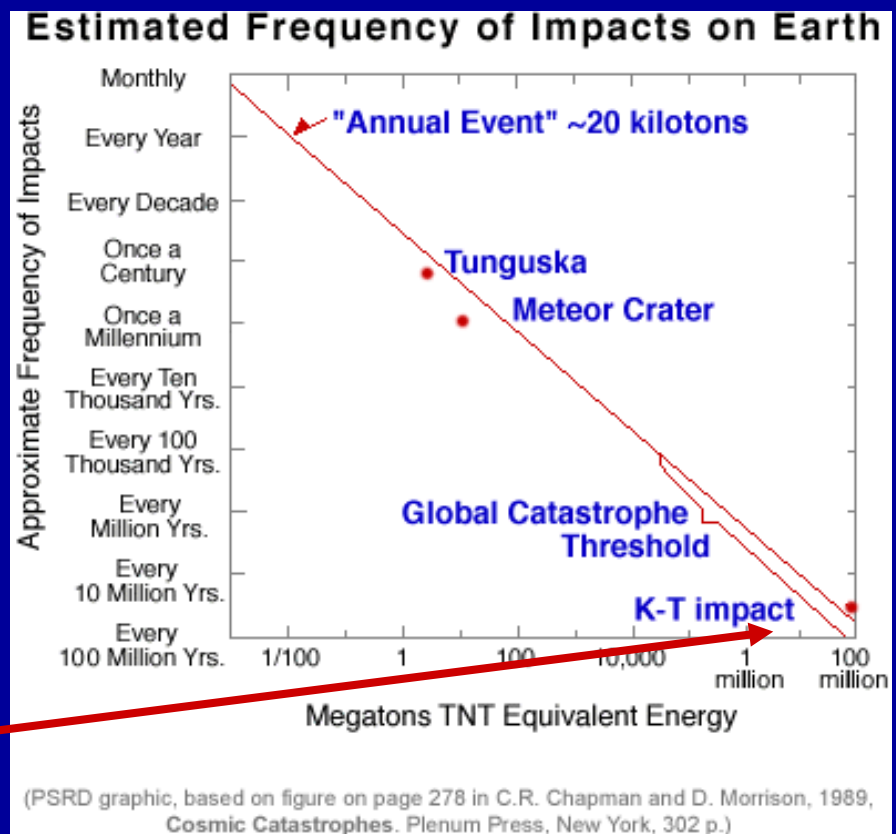




## Meteor Crater in Arizona

The biosphere is not disconnected from what comes in from above, so we might want to think of there being another sphere we might call the Ouranosphere after the Greek word (pre-Christian) for "sky, heaven".

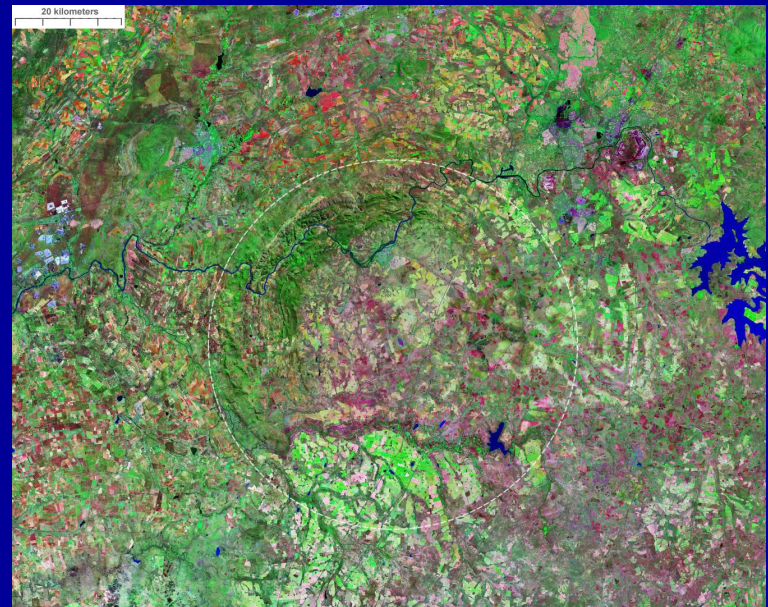
6-8 between 2-4 billion years ago





**Sudbury Crater,  
Ontario, Canada  
1.85 Gya  
155 miles diameter**

**Verdefort Crater,  
South Africa, 2 Gya  
112 miles diameter**



**Chicxulub, 6-9 miles diameter, 66 MYa  
What about those that hit in the oceans? a giant tsunami**





**Crater Lakes are more common than thought.**

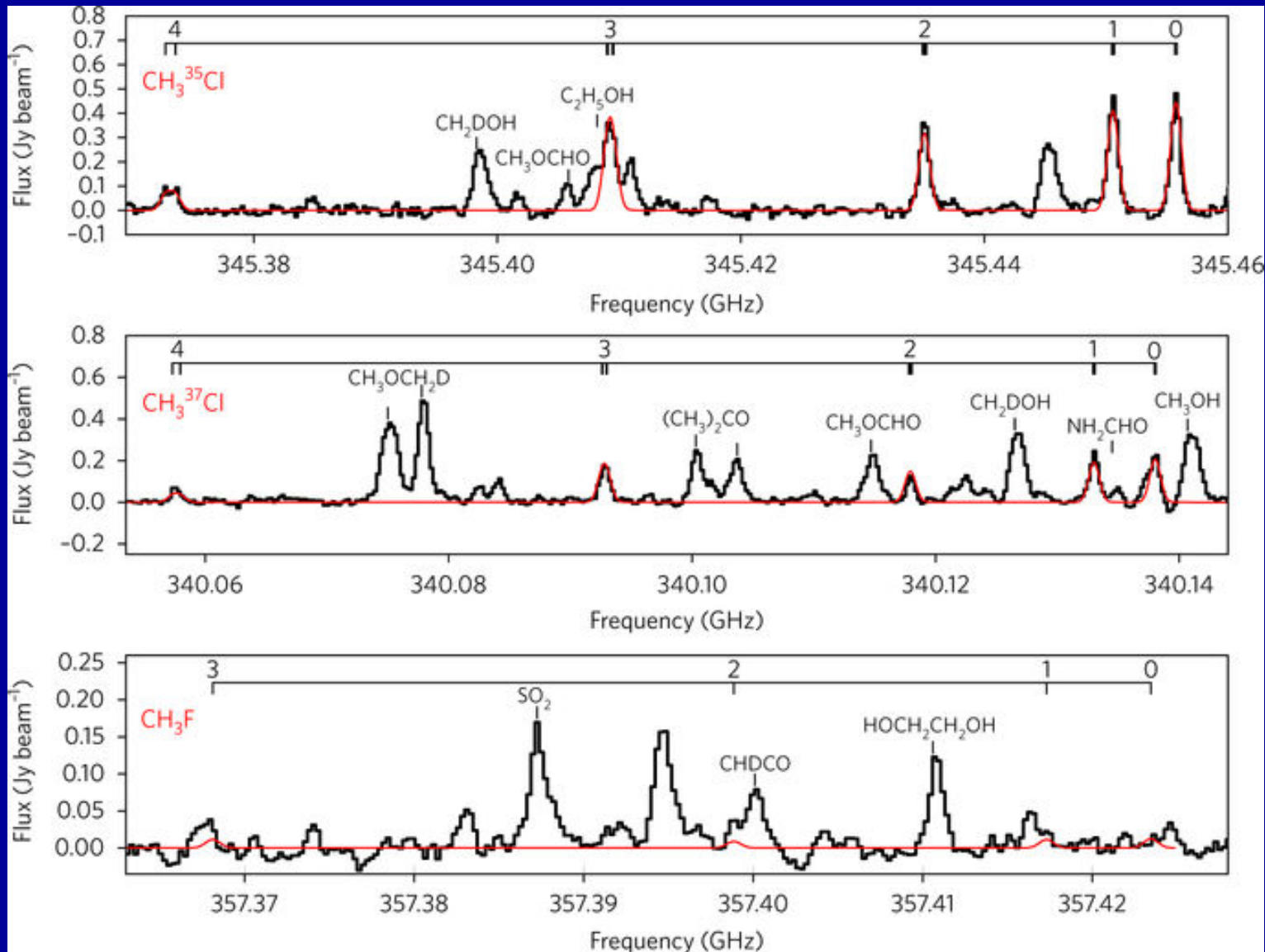


Pingualuit Impact Crater,  
Quebec, Canada



# Far IR spectrum around protostar

300 GHz  $\leftrightarrow$  1 millimeter

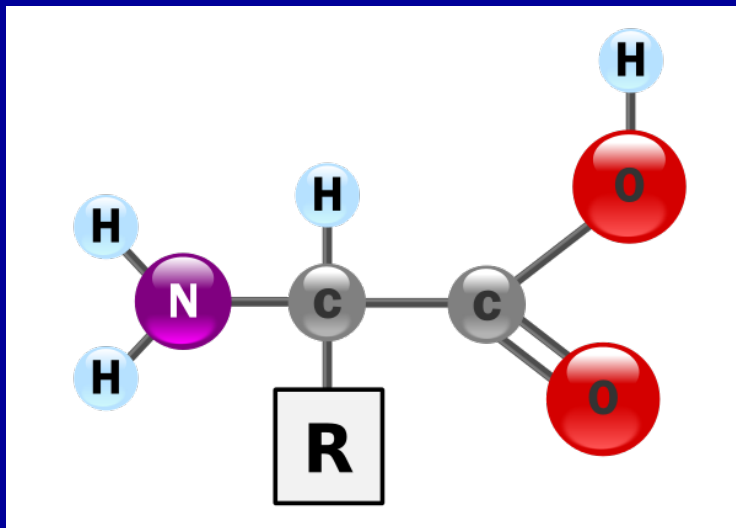




Meteors bring simple organic molecules  
Amino acids, hydrogen cyanide, etc.

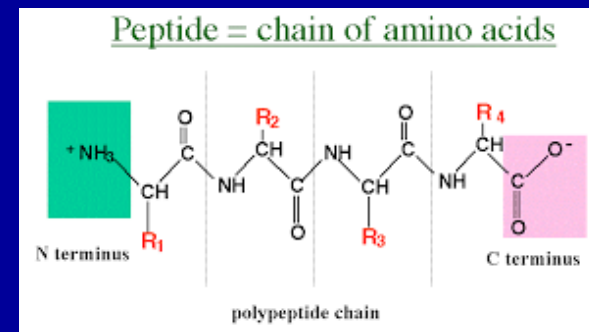


**Amine:  $-NH_2$       carboxyl:  $=COOH$**



**Side group "R" of atoms specific to each acid**

**Collisional energy of "big guys"  
Makes peptides**



# Life on planet Earth

- Easy or hard?
- Same here as on other planets, or fundamentally different?
  - Does life always require water?
- Chance vs. inevitable
- Extremophiles



# Origin of Life

- When?
- Where?
- How?

# Number of events leading to life

$$\langle N_{abiogenesis}(t) \rangle = N_b \cdot \frac{1}{\langle n_0 \rangle} \cdot f_c \cdot P_a \cdot t$$

$N_{abiogenesis}$  : the number of origin events  
probably = 1 or at most a few for Earth

$N_b$  : number of building blocks (must be a big number)

$\langle n_0 \rangle$  : number of building blocks required to get a living system

$f_c$  : the fraction of the available building blocks

$P_a$  : Probability of an origin event to occur ( $10^{-33}??$ ) per unit time  
assembly of the necessary molecules

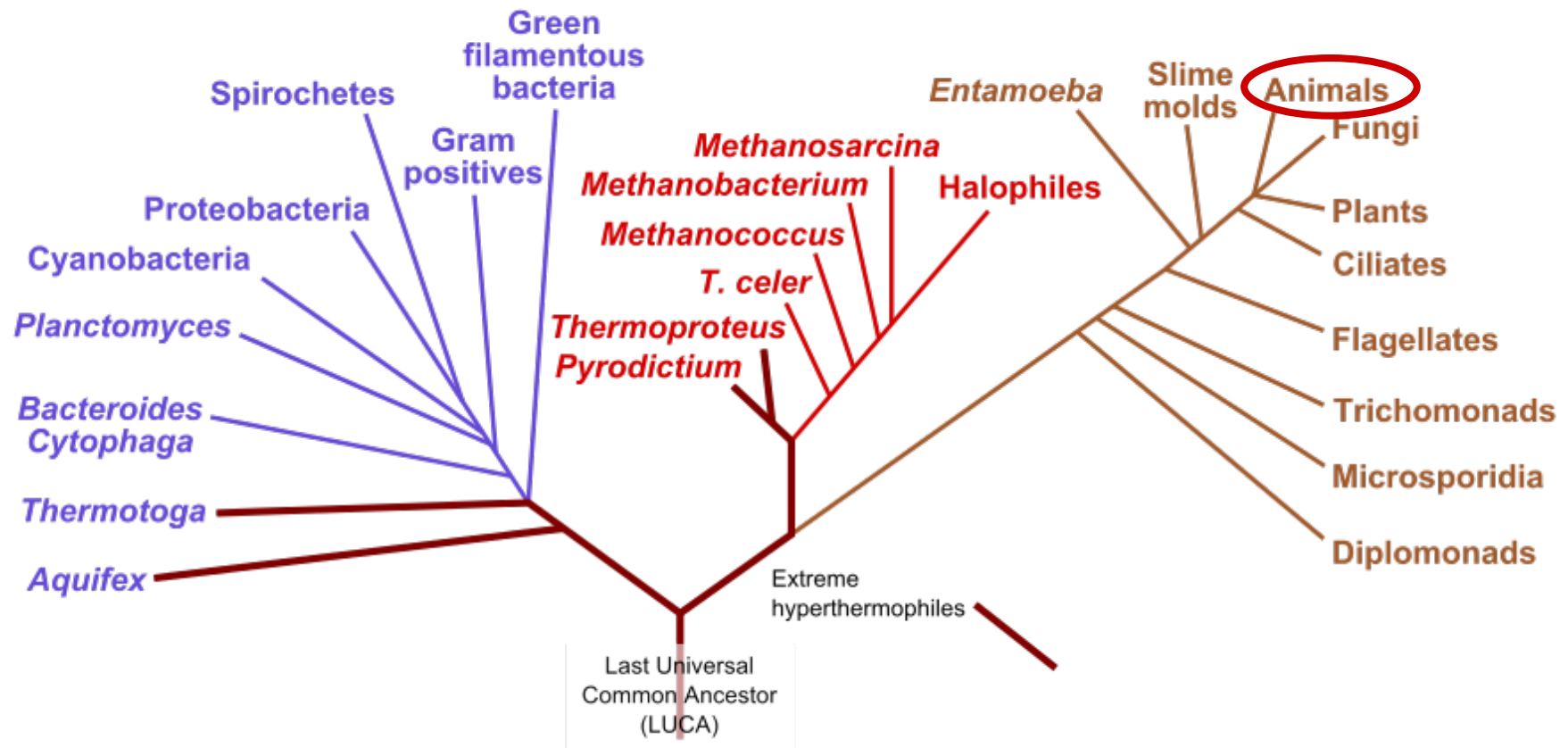
$t$  : time available (100 M yr)

# Phylogenetic Tree of Life

## Bacteria

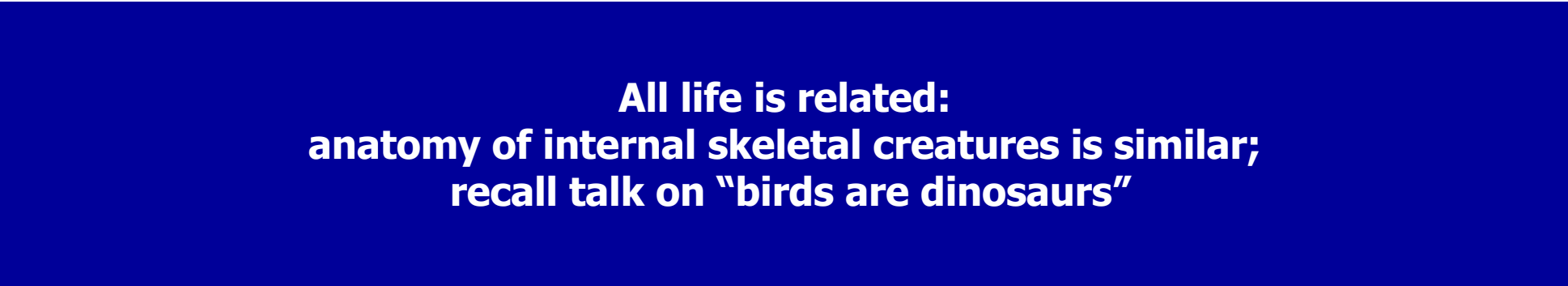
## Archaea

## Eukaryota





# Tree of Life



**All life is related:  
anatomy of internal skeletal creatures is similar;  
recall talk on "birds are dinosaurs"**

# Caveats on tree

1. From mammalian point of view
2. Bacteria hold more genetic diversity
3. Genetic diversity: more branches extinct than exist today
4. Single cell eukaryotes (aka protists) vastly underrepresented
5. Birds not a separate branch – they are a branch of theropod dinosaurs

# Some new thinking

- Life is a basic planetary process
- Living systems have radically altered global planetary chemistry
- Living systems are ordered – many ways and at many levels
  - Entropy; include environment
- Although living systems are diverse and heterogeneous, living order is highly selected
- A simple foundation underlies the unlimited complexity
- Durable feature of Earth



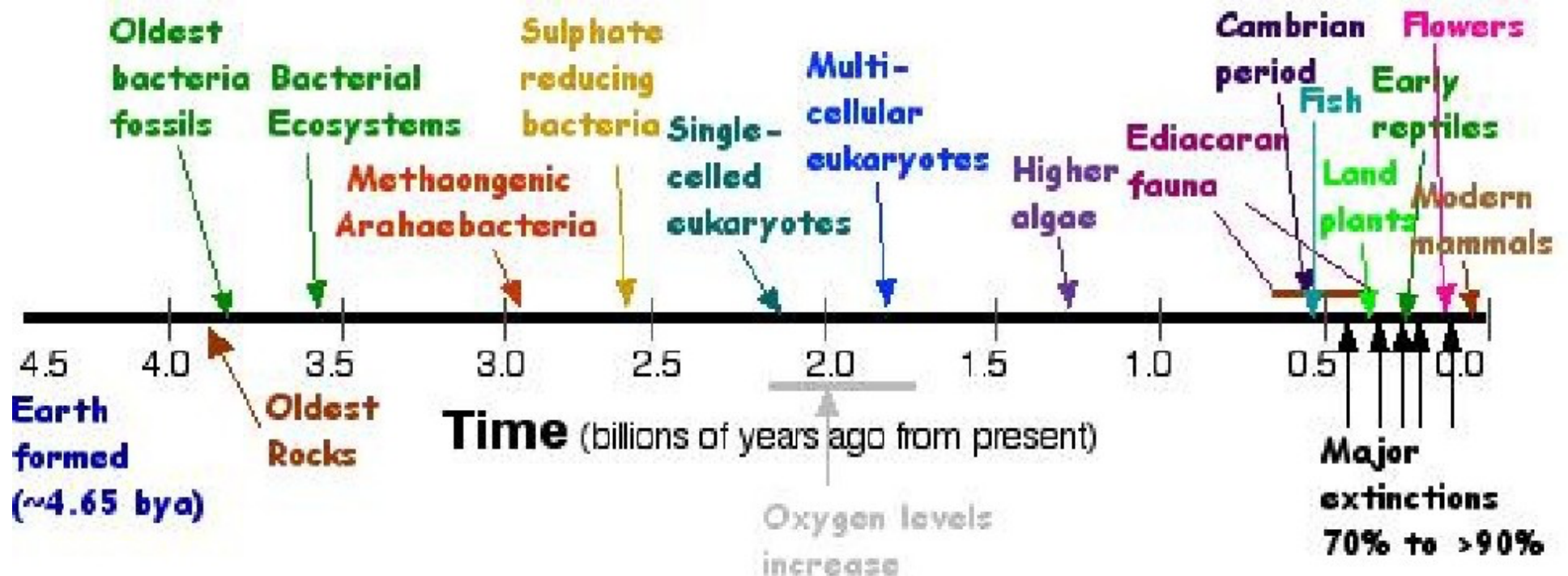
# Interfaces between geospheres

- Chemicals can pass between geospheres; the interfaces can be concentrating centers for thermodynamic disequilibrium and the emergence of complexity.
- Examples:
  - Volcanic activity releases gases into water and air
  - Weathering removes chemicals from rock and transfers to water

Oldest sedimentary rock surface  
showing evidence of bacterial mats  
(3.48 Gya), Pilbara region of Australia.



# Timeline



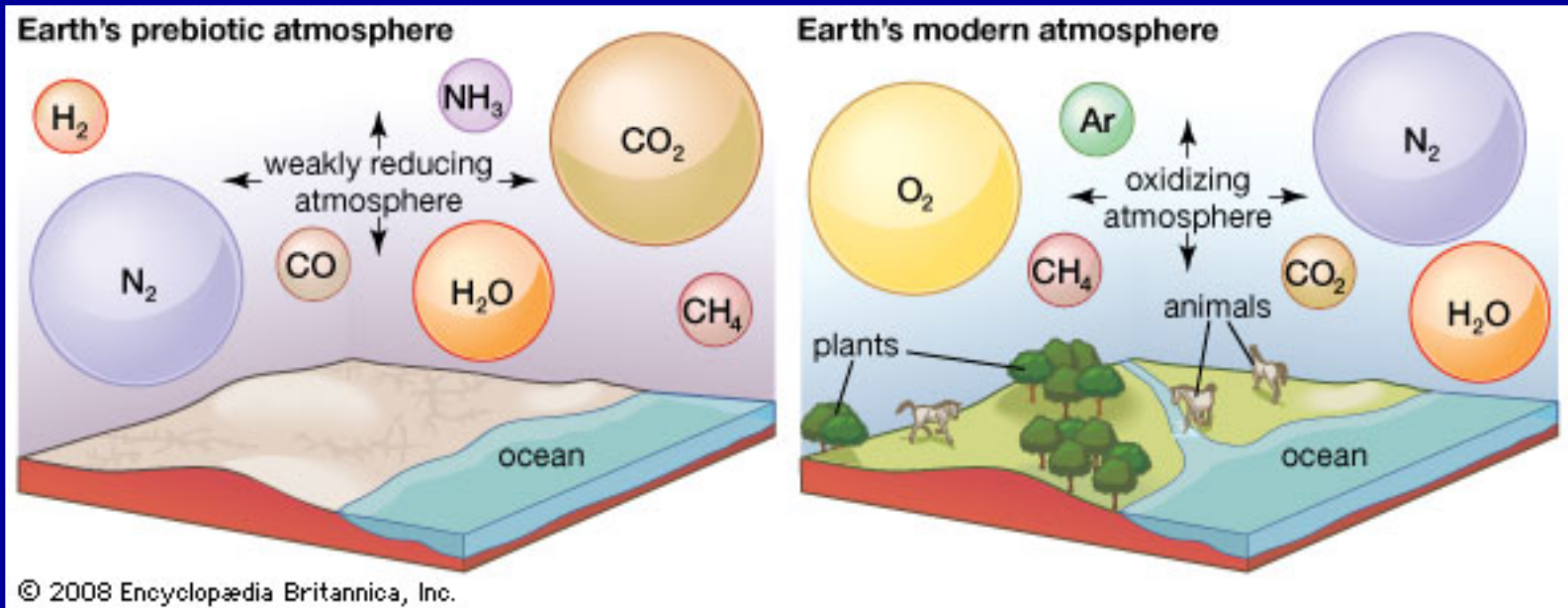
Methanogens still around today as extremophiles found deep in ice, underground and in hot desert environments. Some have a constant metabolism able to repair macromolecular damage, at temperatures of 145 to  $-40^{\circ}\text{C}$ .



**Biota have changed the atmosphere.**

**Oxygenation of the atmosphere by cyanobacteria,  
(anaerobic bacteria), the first to produce oxygen by photosynthesis.**

**began**



# Earth colored: early

- Red
  - iron
- Grey
  - igneous rocks - granite: quartz and feldspar like Yosemite
- Blue
  - water, brought by comets and asteroids
  - Probably enough water came with the original formation of the planet to provide for early forms of life

# Colored Earth: late

- Cyan
  - Cyanobacteria
  - Prokaryote that use sunlight (first photosynthesis)
- Red
  - Rusty rocks
  - Iron oxide hematite
  - Corresponds to the appearance of oxygen
- Green
  - Evolution of plants



# Mineral evolution

- Meteorites probably contain about 50
  - None with oxygen
- Gravity differentiated the elements
  - Liquid Earth
  - Iron fell to the center
  - Magnesium, aluminum, silicon etc. rose
  - As Earth cooled, more minerals formed
- We now have list of about 3500 minerals
  - Many with oxygen



**Color  
changing  
LED base!**  
¡Base LED que  
cambia de color!

**Grow 6  
crystal colors!**  
¡Haz crecer 6 cristales de  
distintos colores!



**Grow crystal dusters!**  
¡Haz crecer racimos de cristales!



**WARNING:** Do not eat or drink from the container. Do not use the container for food or drink. Do not use the container for storage. Do not use the container for disposal.

**CAUTION:** Do not use the kit for other purposes.

**WARNING:** Do not use the kit for other purposes.

**TOTAL NET WT. OF CONTENTS: 17.6 OZ (500 g)**

**Deluxe  
Crystal Growing Kit**  
Kit de cultivo de cristales deluxe

x1  
187 (52.45)  
93 (26.31)



# Miller – Urey

## University of Chicago, early 1950s

<https://www.youtube.com/watch?v=xyhZcEY5PCQ>

**Start 18:33 end 21:35**



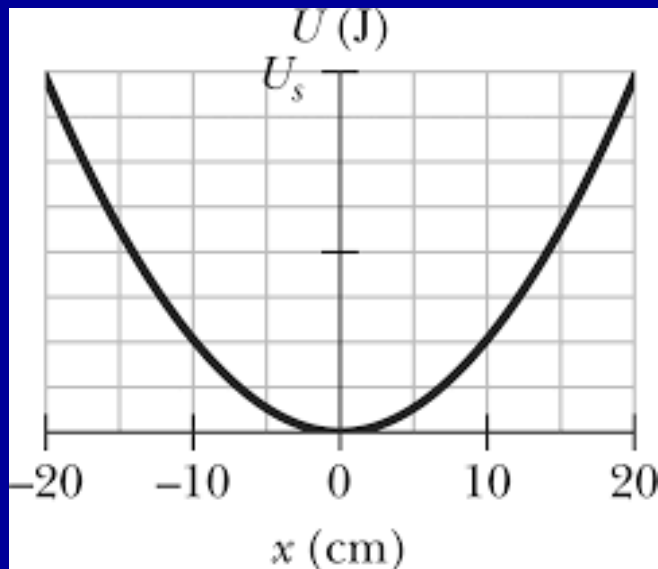
# What is Life?

- Physics and chemistry
  - Matter, forces, energy, entropy, reactions
- Biology
  - Coding, signals, instructions, translation
- Information – anything that could be different
  - How does information come about

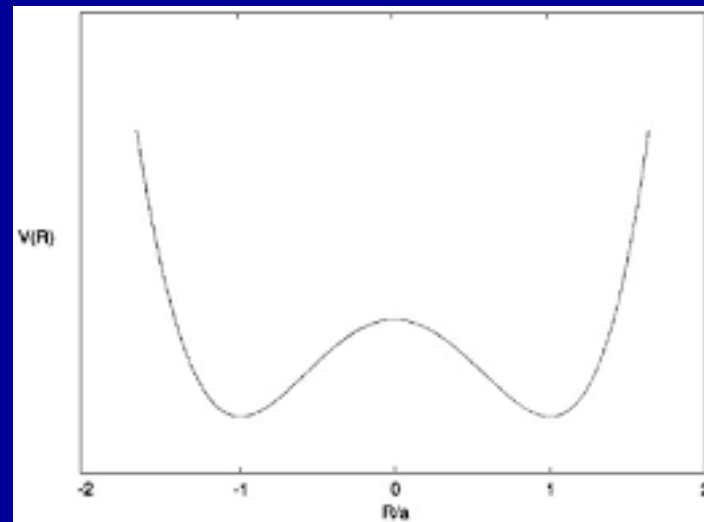
**Stuff -> bits**

**Not to be confused with meaning,  
something given to information by us.**

# An example of information



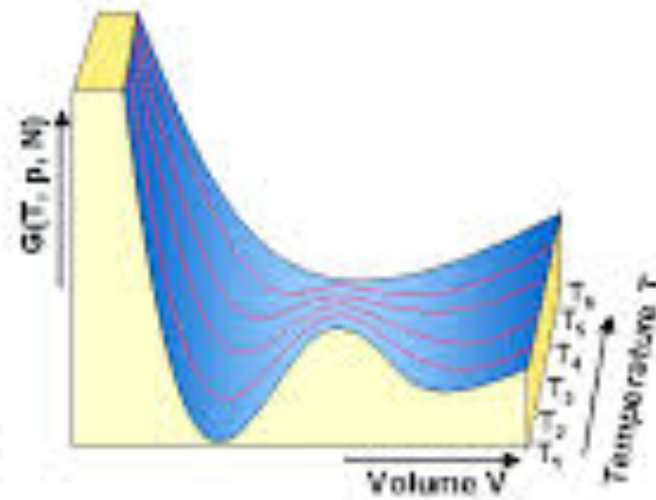
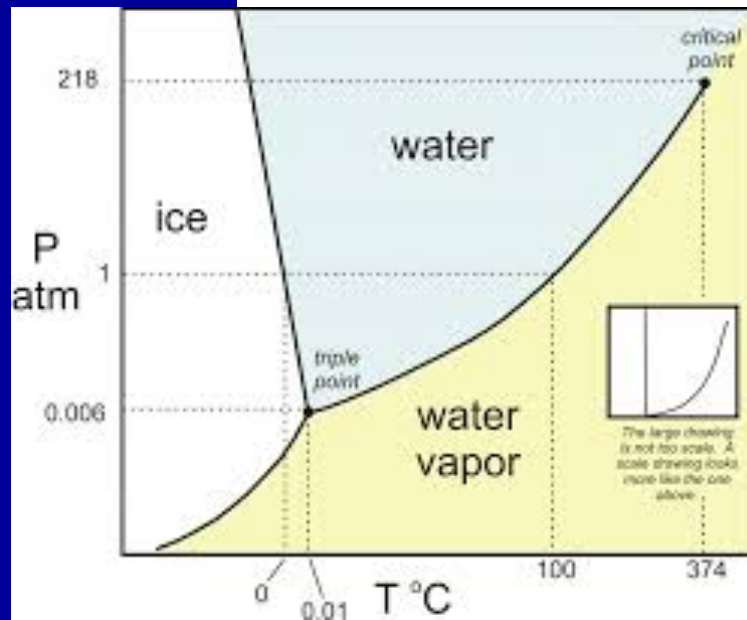
**Stable potential**



**Bistable potential**

# Gibbs function (thermodynamics)

## Second order phase transition



**Phase transitions are probably key to understanding the origin of life**

# Let's get started

**Did you get the email and assignment, and did you look at this short video?**

**<https://www.youtube.com/watch?v=fgQLyqWaCbA>**

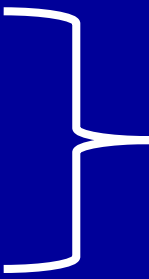
## **Life requires**

- **Metabolic process**
- **Membrane (to make cell walls): lipids**
- **Genetic information: at least RNA, maybe DNA**

**Most definitions include homeostasis: the ability to stay in equilibrium with environment**



# Ingredients to make life

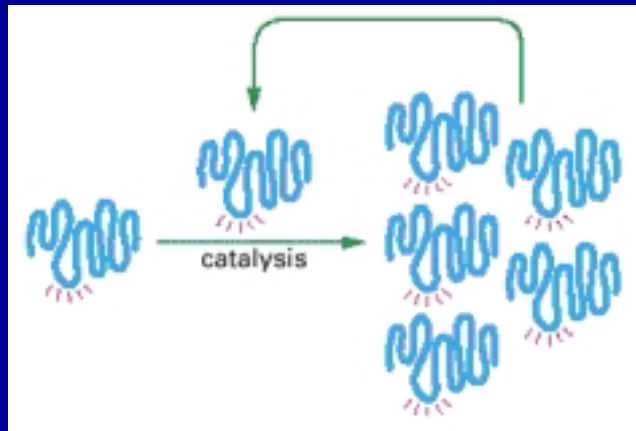
- Carbon
    - Makes chain molecules - complexity
  - Hydrogen
  - Nitrogen
  - Oxygen
- 
- All bond easily with carbon**
- Sulfur (sometimes selenium)
    - Electron shuffle
  - Phosphorous (rarely arsenic)
    - Energy storage

# Abiogenesis

- The evolution of life or living organisms from inorganic or inanimate substances
- Need the right environment
  - Concentrate chemicals
- Much work is going on to find the right environmental conditions and ingredients to make this happen

# Life require self catalysis

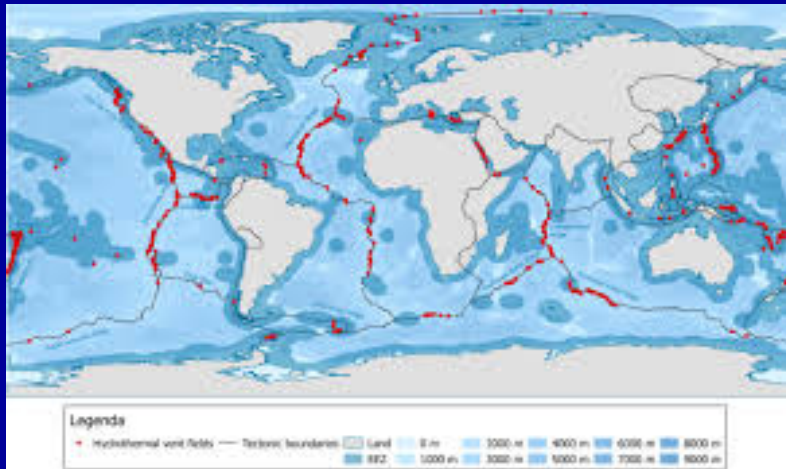
**An hypothetical RNA molecule that can catalyze its own synthesis.**



Molecular Biology of the Cell. 4th edition.  
Alberts B, Johnson A, Lewis J, et al.  
New York: [Garland Science](https://www.ncbi.nlm.nih.gov/pub/books/bk10000/); 2002.

<https://www.ncbi.nlm.nih.gov/books/NBK26876/>

# Hydrothermal vents





# Vent inorganic chemistry

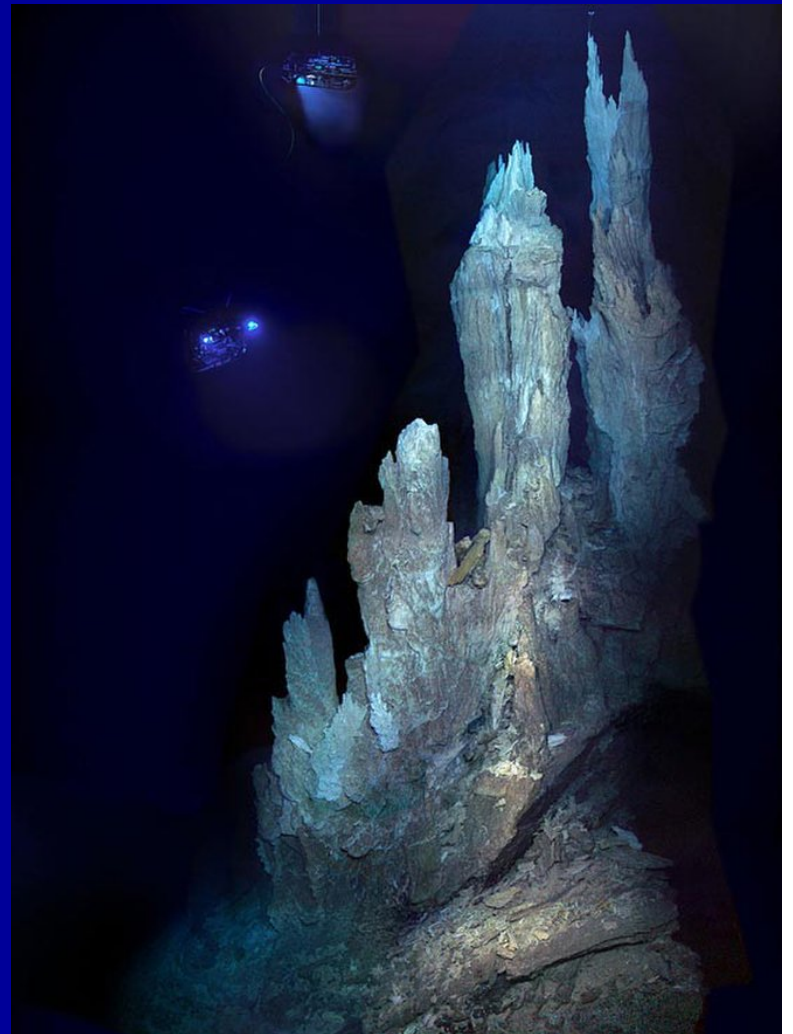
Cations (positive ions) from seawater ( $\text{Mg}^{+2}$ ,  $\text{Ca}^{+2}$ , and  $\text{Na}^{+}$ ) form hydroxyl (OH) bearing minerals in the volcanic rocks, altering the minerals and releasing hydrogen ions to solution. The hot, acidic altered-seawater releases metals (Fe, Mn, Zn, and Cu) and reduced sulfur ( $\text{H}_2\text{S}$ ) from the volcanic rock; these are transported by hydrothermal solutions to the seafloor and form metallic mineral deposits.

Organic compounds can use the ions to catalyze chemical reactions. Hydroxyl-bearing alteration minerals are created.

# Source of dissolved simple organic molecules

High temperatures suddenly cooled by contact with seawater leads to a big change in the equilibrium concentration known as organic quenching. This creates a high concentration and provides a chemical potential energy that allows more complex molecules in the presence of the minerals in the rocks around the geothermal vents.

Called the *Iron-Sulfur World Theory*



# Chemosynthesis

Chemosynthesis uses inorganic molecules (such as hydrogen sulfide) or methane and combines them with an oxygen source (in this case seawater) to create simple sugars.

## Issues:

- How is a supply of “chemical food” for reactions concentrated?
- How do the more complex molecules survive at such high temperatures?

# Photosynthesis vs. chemosynthesis

All photosynthetic organisms use solar energy to turn carbon dioxide and water into sugar and oxygen. There is only one photosynthetic formula:



Photosynthesis occurs in plants and some bacteria, wherever there is sufficient sunlight - on land, in shallow water, even inside and below clear ice.

All chemosynthetic organisms use the energy released by chemical reactions to make a sugar, but different species use different pathways. Some vent bacteria oxidize hydrogen sulfide, add carbon dioxide and oxygen, and produce sugar, sulfur, and water:





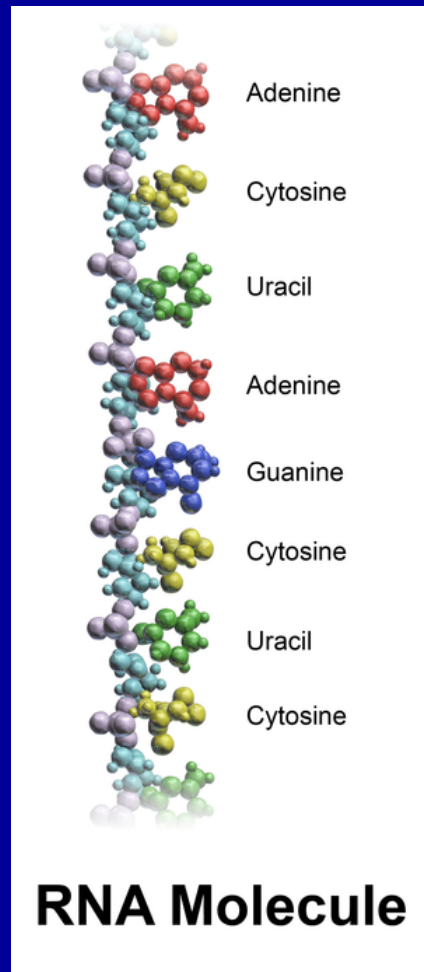
# Minerals may play a big role

- It has been proposed that the first “biological” molecules on Earth were formed by metal-based catalysis on the crystalline surfaces of minerals. An elaborate system of molecular synthesis and breakdown (metabolism) could have existed before the first cells arose.
- Life requires molecules that possess the ability to catalyze reactions that lead, directly or indirectly, to the production of more molecules like themselves.
- Catalysts with this special self-promoting property can use raw materials to reproduce themselves and thereby divert these same materials from the production of other substances.

# Autocatalytic molecules

- In 1982, it was discovered that RNA molecules themselves can act as catalysts and also carry the information required to make proteins.
- This forms the basis of the RNA world hypothesis.
- Although RNA seems well suited to form the basis for a self-replicating set of biochemical catalysts, it is unlikely that RNA was the first kind of molecule to do so. (Difficulties exist!)
- The first molecules to possess both catalytic activity and information storage capabilities may have been polymers that resemble RNA but are chemically simpler.

# Laboratory synthesis of RNA



Short chains of RNA have been synthesized in the laboratory but it seems to be hard to get sufficiently long chains.

*Bridging the prebiotic and RNA worlds: prebiotic RNA synthesis on clay.* Ertem et al. (2000)

- Montmorillonite - a primitive clay
- Have to "feed it" the monomers (A, C, G, and U)

# How to concentrate the building blocks

- Ocean geothermal vents – problem dissipation by ocean currents
- New proposal hot-dry geothermal pools
  - Scientific American, Aug. 2017, p28



# Wet-Dry cycle for the origin of life

- Stable crust of Earth peppered with volcanic landmasses (Yellowstone-like environments common)
- Each with 100 geothermal springs, varying chemical and environmental conditions, energy
- -> millions of combinations of chemicals per year
- 10 millions years – lots of experiments to make life

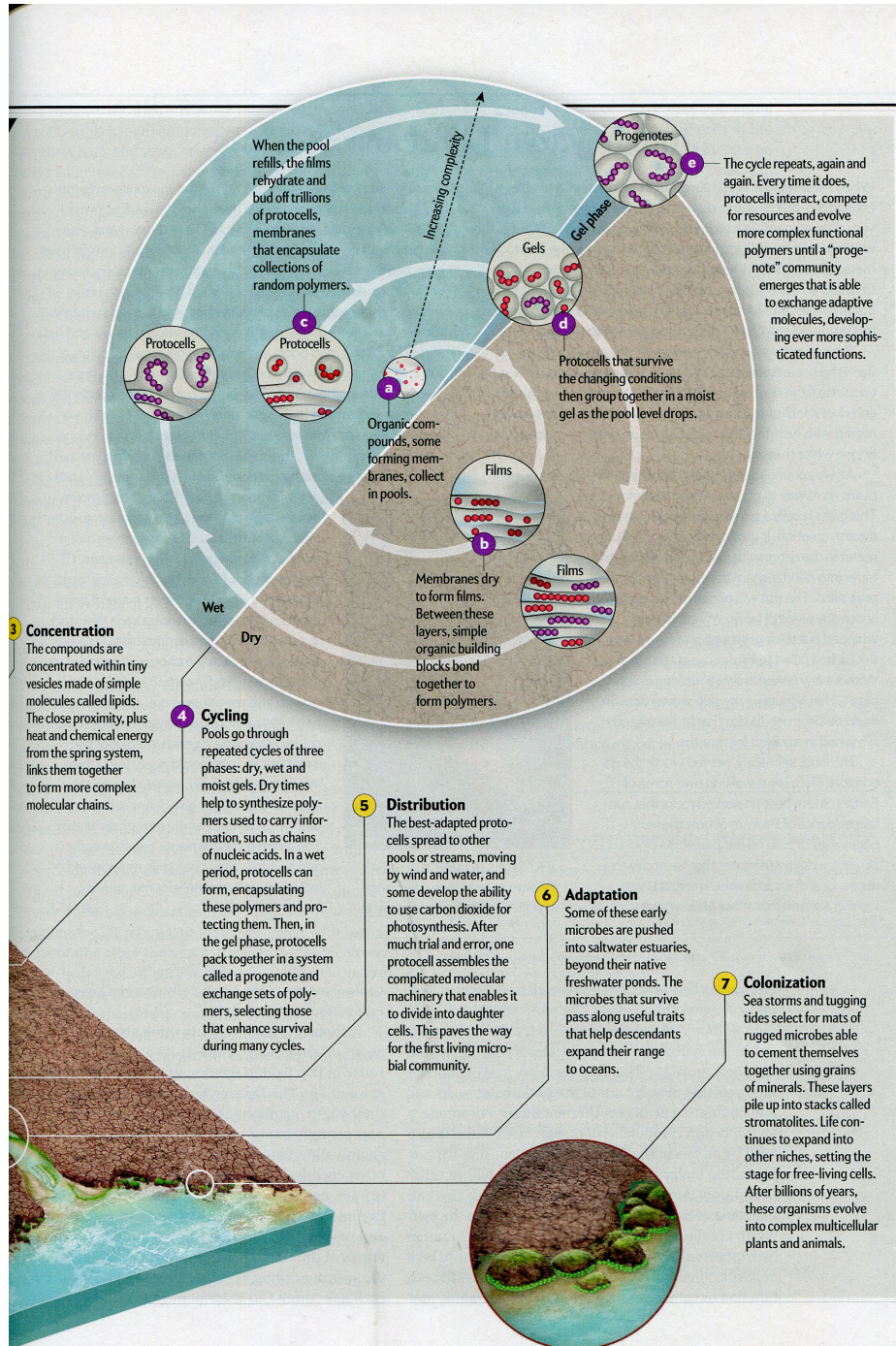


Illustration by José Miguel Mayo (landscape) and Jen Christiansen (cycling detail)



# Geothermal vents: Lake Yellowstone (hot/cool)



# Wet-dry pools Yellowstone









# Throw in some geysers



To help produce wet dry cycling, and (some say) bingo  
– the potential to make a self-replicating molecule.

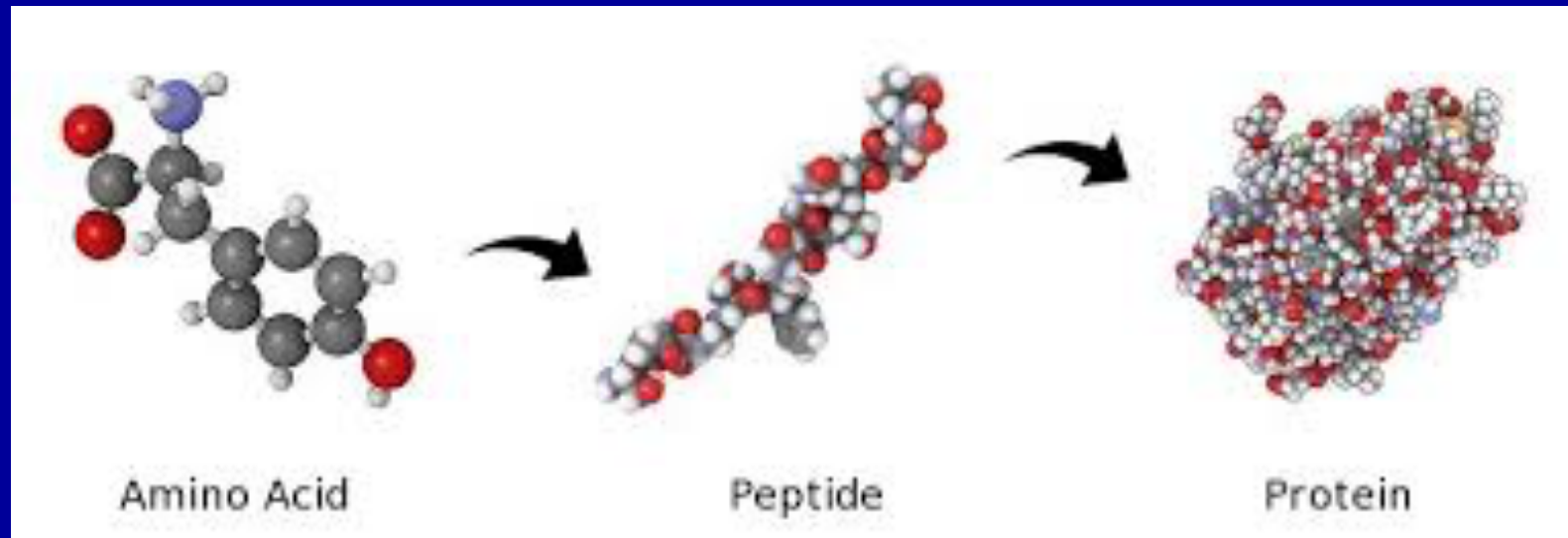
# Seven ideas being explored

1. Lightning in a reducing atmosphere
  - Early Earth atmosphere wasn't reducing, but maybe around volcanos
2. Molecular meetup on clay
  - Maybe as part of a hot-dry model
3. Deep sea vents – concentration in cavities of the rocks
4. Ice protected complex molecules from ultraviolet light.

# Seven ideas

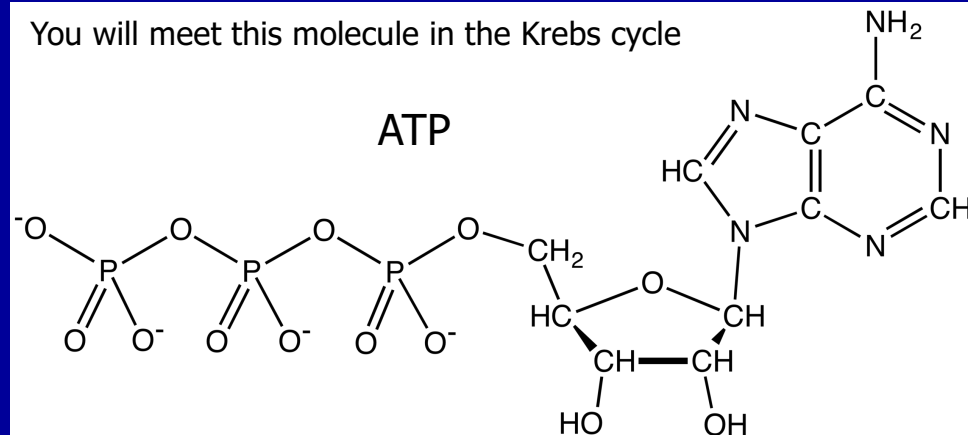
5. DNA formed from RNA, but how did we get RNA? Known as the gene-first hypothesis.
6. Simple molecules did the reaction but were replaced by more complex but more efficient processes by more complex molecules – metabolism first
7. Panspermia – life came here from somewhere else.

# How to get long chains of peptides that might form RNA?



**Key is  
adenosine triphosphate  
aka ATP**

You will meet this molecule in the Krebs cycle





# Stromatolites



**Live stromatolites in western Australia**

**cyanobacteria**

**Blue-green algae that learned how to photosynthesize sunlight and produce oxygen as a byproduct, eventually building up oxygen in the atmosphere.**

**Reacted with abundant iron in oceans to produce red rocks (rusted the planet).**



**Fossilized stromatolites found with oldest rock on Earth in Australia**

# Christian de Duve 1917 - 2013

**"If you equate the probability of the birth of a bacteria cell to chance assembly of its atoms, eternity will not suffice to produce one... Faced with the enormous sum of lucky draws behind the success of the evolutionary game, one may legitimately wonder to what extent this success is actually written into the fabric of the universe".**



de Duve lecturing on the origin of the eukaryotic cell in October 2012

# How Life Began Origins Nova Neil Degrasse Tyson

<https://www.youtube.com/watch?v=NJQ4r81DZtY>

**Start 4:47**

**End 50:49**